Assessment of Heavy Metals in Water, Fish and Sediments from UKE Stream, *Nasarawa* State, Nigeria

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ABSTRACT

The levels of lead, zinc, copper, iron, manganese, cadmium and mercury were determined in various body parts of two species of catfish; Clarias gariepinus and Synodontis schall, water and sediment samples from Uke stream using atomic absorption spectrophotometer (AAS) method. The results obtained showed that iron (Fe) had the highest concentration with average of 8.78 mg/g and 7.51 mg/l in sediment and water respectively followed by Zn with 4.79 mg/g (sediment) and 3.19 mg/l (water) while Cd had the lowest concentration of 0.035 mg/g and 0.023 mg/l in the sediment and water respectively. In the two fish species, zinc (0.17 - 3.25 mg/g) was the most highly concentrated in the various matrices while lead (0.011 - 0.031mg/g) was the lowest. Metal levels in the various body parts of the two species of fish studied were found to be more concentrated in either, the head, gills or the intestine. In both species zinc had the widest variability while lead was the least. The metal levels determined in water and sediment are all above the tolerable limits recommended by regulatory bodies which is an indication that this ecosystem is contaminated with heavy metals which would eventually end up in the food chain. The metals determined in various body parts of two species of catfish were below deleterious level; however there is the need for regular monitoring of the heavy metal load in this water body and the aquatic organisms in there because of the long term effects.

Key words: Clarias gariepinu, Synodontis schall, water, sediments, heavy metals, AAS.

INTRODUCTION

Over the last few decades, there has been growing interest in determining heavy metal levels in the marine environment and attention was drawn to the measurement of contamination levels in public food supplied, particularly fish¹-³. Although heavy metal is a loosely defined term⁴, it is widely recognized and usually applied to the wide spread contaminants of terrestrial and fresh water ecosystems. Some examples of heavy metal include lead, zinc, cadmium, copper and manganese. Many of these heavy metals are toxic to organisms at low concentration⁵-6.

The concentration of metals in bio-available form is not necessarily proportional to the total concentration of the metal. The concentration of various elements in the air, water and land may be increased beyond their natural level due to the agricultural, domestic and industrial effluents. These substances are described as "contaminants" when discharged to the environment. In water, insoluble heavy metals may be bound to small silt particles. Metals and other fluvial contaminants in suspension or solution, do simply flow down the stream, they form complexes with other compounds settle to the bottom and ingested by plants and animals or adsorbed to sediments. Consequently, aquatic organisms may acquire heavy metals in body

directly from the water via gills or food chain mechanisms⁹.

Work has been presented on heavy metal concentrations in water, sediments and fishes from Nasarawa and Antau streams in Nasarawa State to ascertain the extent of heavy metal pollution in these aquatic ecosystems and their eventual uptake by aquatic organisms. Different parts of fish (head, gills, intestine and flesh) were used and by drying these parts of fish and the sediment samples and employing the method of wet digestion and AAS, the level of metals in these parts of fish and the sediment were determined. The level of metals in water was also determined using AAS after pretreatment. The results showed the presence of metals determined in all the samples but were below the deleterious level^{5,10}.

Water and sediments are commonly used as indicators for the state of pollution of aquatic ecosystem¹¹. Uke stream runs through the centre of the town and the water from this stream serves domestic purposes as well as irrigation farming and aquatic organisms (fish) from this water body is one the major sources of protein for the populace of this location. However, the stream also serves as points of discharge for domestic wastes in some areas along the length of the stream and runoffs from agricultural lands always flow into the stream at different points.

Aquatic animals (including fish) bioaccumulate heavy metals in considerable amount in tissues over a long time and the dependence of the populace in this area on this water body for domestic water supply and its aquatic organisms (fish) as source of protein makes it imperative to assess the level of heavy metals in this aquatic ecosystem in view of the health implications that cut across the food strata.

This research reports the level of Pb, Zn, Fe, Cd, Cu and Mn in parts of fish caught from Uke stream as well as the stream water and sediments in order to ascertain the relationship between bioaccumulation of these metals in the aquatic organisms (fish) with the distribution and concentrations of these metals in the stream water and sediments. This is aimed at ensuring the safety

of this ecosystem for the benefits of the residents of Uke and its environs.

MATERIALS AND METHODS

Collection of samples

Water samples were collected using plastic containers to fetch water below the surface at designated points, mixed properly and stored in a plastic container rinsed with 0.01N nitric acid and kept in deep freezer prior to the time of analysis11. The sediment samples were collected by scooping with a plastic spoon from the points where the water samples were taken, air dried and kept awaiting analysis. The samples of available fish species (Clarias gariepinus and Synodontis schall) in the stream were purchased from fishermen at the stream site. They were properly and carefully washed and stored at 4°C pending analysis. These samples were all collected at 7.00 local time while the temperature (28°C) of the water was taken at the point of collection.

Sample treatment

Five (5.0) cm³ of concentrated hydrochloric acid were added to 250.0 cm3 of water sample and evaporated to 25.0 cm³. The concentrate was transferred to a 50.0 cm³ standard flask and diluted to the mark with de-ionized water11. 5.0 g of prepared sediment sample was digested with 15.0 cm³ nitric acid, 20.0 cm3 perchloric acid and 15.0 cm3 hydrofluoric acid and placed on a hot plate for 3h. On cooling, the digest was filtered into a 100.0 cm³ volumetric flask and made up to the mark with distilled water¹². Different body parts of the fish (Head, gills, intestine and flesh) were dried in the oven at 105°C until constant weight is obtained and blended. 2.0 g of the blended fish parts were weighed and digested using the approved method14.

Mineral analysis

Lead, zinc, copper, iron, manganese, cadmium and mercury were determined in samples of fish body parts of *Clarias gariepinus* and *Synodontis schall*, water and sediment using computer controlled Atomic Absorption Spectrophotometer (AAS VGB 210 System). The instrument setting and operational conditions were done in accordance with the manufacturers'

specifications. All determinations were in triplicates.

Statistical analysis

The results obtained were subjected to statistical evaluation. Parameters evaluated were grand mean, standard deviation (SD) and coefficient of variation (CV %).

RESULTS AND DISCUSSION

Table 1 shows the mean metal concentrations, grand mean, standard deviation and coefficient of variation percent of water and sediments. The mean concentration of metals determined in the water samples ranged from 0.023 – 7.51 mg/L and for sediments the range was 0.095 – 8.78 mg/g. The metals determined were Pb, Zn, Fe, Cd, Cu, Mn and Hg with mean concentrations of 0.040, 3.19, 7.51, 0.023, 0.95, 0.51 (mg/l) in water and 0.095, 4.79, 8.78, 0.035, 1.34, 0.24, and (mg/g) in sediment respectively with Hg not detected in both water and sediment samples. The concentrations of Fe, Zn, in both samples and Cu in sediment were high (> 1.0 mg/L). The concentration of Fe being highest in sediment

agrees with the result of the report of heavy metals in sediments of Rafin Mallam stream¹⁰ but its value in water, also highest is high than the value recorded in the report of heavy metals in water and fish from River Antau⁵. However, the concentration of Fe in sediments and water to an extent is determined by the nature of soil along the stream banks15 from where it is leached into the water body and sediments. The values of Zn recorded are lower than the ones obtained from the results of the report of heavy metals in water, sediments and fish from River Nasarawa and for Cu, its concentration is the same for water and higher for sediment than that obtained for River Nasarawa¹⁰. Zinc is widely used for making paints, dyes, rubber, wood preservatives and through wares and tears; zinc from this sources is discharged into the environment. Although zinc is required by plants and animals for normal growth, higher concentration of it is toxic¹⁶.

The concentrations of Cd and Pb obtained are lower for water and higher for sediments from the work on a water body used for irrigation in Keffi¹⁰. The concentration of Mn in sediment is lower and higher in water than those obtained from the work

Table 1: Mean metal concentrations in water (mg/L) and sediment (mg/g)

Parameter	Pb	Zn	Fe	Cd	Cu	Mn	Hg
Water	0.040	3.19	7.51	0.023	0.95	0.51	ND
Sediment	0.095	4.79	8.78	0.035	1.34	0.24	ND
Grand mean	0.068	3.99	8.15	0.029	1.15	0.38	-
SD	0.039	1.13	0.90	0.008	0.28	0.19	-
CV%	57.35	28.35	11.04	2.76	24.35	50.00	-

standard deviation, SD; coefficient of variation percent, CV%.

Table 2:National and International Standards

Metals	Wat	er (mg/L)	Fish	(mg/g)	
	FDA	WHO	EPA	WHO	
Pb	0.005	0.01	0.05	1.5	
Zn	-	3.0	5.0	150	
Cu	1.0	1-2	1.0	-	
Fe	-	0.3	0.1	-	
Mn	-	0.1-0.5	0.05	2.5	
Cd	0.005	0.003	-	0.2	

carried out on River Tammah¹². Cu, Pb and Mn are some of the metals that get into aquatic ecosystem from runoffs from agricultural lands as a result of the use of agrochemicals containing heavy metals such as Cu, Mg, Mn, Pb and Zn¹⁷. The probable sources of Cd in surface water includes leaching from Ni – Cd batteries, run off from agricultural soils where phosphate fertilizers are used and other wastes¹⁸⁻¹⁹.

All the metals determined were above the World Health Organization (WHO) safety standards, Food and Drugs Administration (FDA) Table-2, and the United States Environment Protection Agency (USEPA) maxima²⁰ except for copper in water which is within the range recommended by W.H.O higher than the ranges recommended by other regulatory bodies. The level of zinc in sediment is however within the range recommended by EPA²⁰. Iron though high above WHO safety standard, it is still safe because it has benefits to organism though in very high concentration leads to conjunctivitis, chroiditis and retinitis if it is in contact and remains in the tissue but Cd is a toxic metal and has no metabolic benefits to human and aquatic biota²¹. Its presence in any compartment of the aquatic ecosystem indicates contamination.

The high level of these metals in both the water and sediment samples are as a result of the runoffs during the rainy season from agricultural

fields and the dumping of domestic wastes in the water body at different points along the length of the stream as they are known to contain heavy metals such as As, Cd, Co, Cu, Fe, Hg, Mn, Pb, Ni and Zn which will eventually end up in this aquatic ecosystem²³.

Among all these metals determined iron has the highest concentration with average of 8.78 mg/g and 7.51 mg/l in sediment and water, respectively followed by Zn with 4.79 mg/g (Sediment) and 3.19 mg/l (Water) while Cd has the lowest concentration of 0.035 mg/g and 0.023 mg/l in the sediment and water respectively. These results are in agreement with the result of in which Cd was found to have the lowest concentration in both sediment and water. Going by the calculated coefficient of variation percent (CV %) when the levels of metals in water and sediment were compared the variability was highest in lead while cadmium was the least varied.

Table 3: Mean metal concentrations in the body parts of African catfish (Clarias garipienus), mg/g

Parameter	Pb	Zn	Fe	Cd	Cu	Mn	Hg
Head	0.031	0.17	0.13	0.005	0.05	0.22	ND
Gills	0.022	2.35	1.63	0.016	1.31	0.12	ND
Intestine	0.012	0.26	0.26	0.025	0.31	0.21	ND
Flesh	0.013	0.18	0.13	0.001	0.15	0.11	ND
Grand mean	0.020	0.73	0.54	0.012	0.46	0.37	-
S.D	0.009	1.08	0.73	0.011	0.58	0.27	-
CV (%)	45.00	147.94	135.18	91.67	126.08	72.97	-

SD standard deviation; CV% coefficient of variation percent; ND not detected

Table 4: Mean metal concentrations (mg/g) in the body parts of African catfish (Synodontis schall)

Parameter	Pb	Zn	Fe	Cd	Cu	Mn	Hg
Head	0.021	0.19	0.12	0.005	0.06	0.24	ND
Gills	0.014	3.25	1.52	0.015	1.35	0.11	ND
Intestine	0.011	0.21	0.31	0.026	0.25	0.18	ND
Flesh	0.012	0.17	0.16	0.001	0.15	0.11	ND
Grand mean	0.015	0.96	0.53	0.118	0.45	0.39	-
S.D	0.005	1.53	0.67	0.111	0.60	0.26	-
CV (%)	33.33	159.38	126.41	94.06	133.33	66.67	-

SD standard deviation; CV% coefficient of variation percent; ND not detected

Table-3 and Table-4 show mean metal concentrations in the body parts (head, gills, intestine and the flesh) of African catfish, Clarias garipienus and Synodontis schall respectively. The presence of these metals analysed in the body parts of fish serves as an indicator for the extent of heavy metal pollution of the water body from where these aguatic organisms (fish) are obtained¹⁰. Also the presence of most of the metals determined in the fish parts agrees with the results of the report of the level of heavy metals in aquatic organism from different water bodies²⁴⁻²⁵ which showed that aquatic animals, fish, inclusive bio-accumulate heavy metals in considerably amount, and because these metals are not bio-degradable, the metal tend to stay in the fish tissues for a very long time which upon consumption of these fish, the heavy metals get transferred to man, leading to heavy metal poisoning in man especially if present in higher concentrations. In both species of the fish zinc presented the highest concentrations followed by iron in the various parts studied and this is relative to the concentrations of these metals observed in both water and sediment samples.

Copper was the next metal after zinc and iron with concentration range of 0.05-1.35 mg/g in the various parts of both species of the fish studied. Lead, cadmium and manganese showed different distribution among the various parts of fish. Mercury was not detected in any part of both species just as it was neither detected in water nor sediment samples. Zinc showed the highest variability of 147.94 and 159.38 % in *Clarias garipienus* and *Synodontis schall* respectively with lead having the least in both cases. This results agrees with that obtained for the analysis of the levels of metals in organs of *Clarias lazera* from river Nasarawa¹⁰.

Most of the metals had highest concentrations either in the head part (lead and manganese) or the gills part (zinc, iron and copper).

Table 5: Bioconcetration factors of the various metals in the body parts of Clarias garipienus

Parameter	Pb	. Zn	Fe Cd	Cd	Cu	Mn	Hg
							9
Head	0.78	0.05	0.02	0.22	0.05	0.43	-
Gills	0.55	0.74	0.28	0.70	1.38	0.24	-
Intestine	0.30	0.08	0.04	1.09	0.33	0.41	-
Flesh	0.33	0.06	0.02	0.04	0.16	0.22	-
Grand mean	0.49	0.23	0.09	0.51	0.48	0.33	-
S.D	0.22	0.34	0.13	0.48	0.61	0.11	-
CV (%)	44.90	147.82	144.44	94.12	127.08	33.33	-

SD standard deviation; CV% coefficient of variation percent

Table 6: Bioconcentration factors of the various metals in the body parts of Synodontis schall

Parameter	Pb	Zn	Fe	Cd	Cu	Mn	Hg
Head	0.53	0.06	0.02	0.22	0.06	0.47	-
Gills	0.35	1.01	0.20	0.65	1.42	0.22	-
Intestine	0.28	0.07	0.04	1.13	0.26	0.35	-
Flesh	0.30	0.05	0.02	0.04	0.16	0.22	-
Grand mean	0.37	0.30	0.07	0.51	0.48	0.32	-
S.D	0.12	0.47	0.08	0.49	0.64	0.12	-
CV (%)	32.43	156.67	114.29	96.08	133.33	37.50	-

SD standard deviation; CV% coefficient of variation percent

This is as result of the fact that the gills helps in respiration and filtration of water²⁴. Relatively high concentrations of some of the metals were found in the intestine since the intestine is part of the visceral muscles which concentrates toxic metals²⁵. Zinc and copper are mineral elements which are essential metals and play vital role in enzyme activity and iron is very important in heamoglobin formation. Lead and cadmium are toxic at very low concentration and have no known functions in biochemical processes. Sources of cadmium include wastes from cadmium- based batteries, incinerators and runoff from agricultural soils where phosphate fertilizers are used since cadmium is a common impurity in phosphate fertilizers¹⁸. Lead is mainly from storage batteries, type metal and antiknock compound in petrol²⁷. The levels of all the metals determined were however below the concentrations recommended by regulatory bodies²¹.

Table-5 and Tables-6 show the bioconcentration factors for *Clarias garipienus* and *Synodontis schall*, respectively. Most of the values obtained for the various fish parts were relatively low (< 1) which showed there was no biological magnification of metal concentration in fish samples except for cadmium (in the intestine) and copper (in gills) for *Clarias garipienus* and zinc (in gills), cadmium (in the intestine) and copper (in gills) for *Synodontis schall*. Order of bioconcentration in the various body parts of *Clarias garipienus* is Zn > Fe

> Cu > Cd > Pb > Mn while that of Synodontis schall is Zn > Cu > Fe > Cd > Mn > Pb. Zn showed the widest variation in both species of fish while the least variations were recorded in Mn and Pb for $Clarias \ garipienus$ and $Synodontis \ schall$, respectively. The levels of metals in the body parts of the two species of catfish were lower than that of the water or sediment. However the presence of metals in the two fishes biochemically showed that fish is relatively dependent on the levels of metals available in aquatic ecosystem.

CONCLUSION

This research has presented data on the levels of heavy metals in sediments, water and various body parts of two species of catfish from Uke stream in Uke town, Nasarawa State. Although the results obtained does not show any form of danger posed to consumers of sea foods and water from this stream but the possibility of deleterious effects after long period cannot be ruled out. This is as a result of the fact that this water body serves as the receptor for domestic wastes as well as runoff from agricultural lands where phosphate fertilizers and other agrochemicals are frequently used. There is therefore the need for continual assessment of the level of pollution of this stream with metals from the mentioned sources with a view to reducing the level of pollution via education and public enlightenment.

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