

Concentration of heavy metals in water and sediment samples from ERO river in Southwestern, Nigeria

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ABSTRACT

The distribution of heavy metal concentration in water and sediment samples from Ero River in South western, Nigeria was carried out for a period of dry and wet season. For both seasons Ni, Cd, Co were not detected in the water samples, while all metal examined were detected in the sediment samples. The concentrations of heavy metals were higher in the sediments than water sample. On the average the concentration of most the heavy metals appears higher in the dry season than wet season. The result obtained fell within the maximum allowable limit set by United State Environmental Protection Agency and World Health Organization.

Key words: ERO River, heavy metals, concentration, water and sediment.

INTRODUCTION

With the global aspiration for the provision of potable water for the teeming population, management of water resources with respect to storage, presentation, acceptability and distribution should be of utmost important, this is the reason why the integrity of some major rivers that serves as tributaries to our major dams are kept as regards pollutants distribution. The advent of technological advancement and industrial revolution in many parts of the world, has resulted into chemical pollution of some of our water system that has posed serious challenges for human existence-potable of there pollutants are mainly heavy metals.

The introduction of pollutants into the environment including water from dams, rivers, estuaries could either be from natural or artificial sources, the problem of water pollution with heavy metals has become a global issue most especially in developing countries as a result of the phenomenal growth in both population and industrial activities (Ajayi *et al*, 1980; Asaolu *et al.*, 1997). Some of the artificial sources of metal introduction

into the aquatic environment include oil spillage, sewage effluents, auto emission, dredging activities, garbage dumps and industrial activities such as minning, canning e.t.c (Books and Rumsey, 1974: Finery *et al.*, 1990: Ipinmoroti and Oshodi, 1993, Asaolu *et al.*, 1997).

Domestic sewage and urban water runoffs (Under Wood, 1971) have been recognized as source of trace metal pollution of the aquatic ecosystem. Domestic sewage contains household detergents, which are known to contain trace amount of Fe, Mn, Cr, Zn e.t.c. Metals can also enter aquatic environment by the use of plant nutrients like fertilizer and crop protective measures. This occurs on farmland in urban and rural areas (Adefemi *et al.*, 2008). Ero river in Ekiti-State runs through farmlands where rock phosphate and phosphates fertilizer which contain high levels of trace metal especially Cadmium (Forstner and Wittman, 1979) are been used. The use of herbicides as gross control measure along highways and other areas have given rise to concern as some of these herbicides contain trace metals. The objective of the present study is to assess the level of heavy

metals in both the water and sediments of Ero River located in Southwestern part of Nigeria. Ero River is one of the river that flows to Ero dam which is one of the major dams in Ekiti State that provide potable water to the teaming population in some area of the State. The results obtained will reveal the levels of heavy metals in the river and its possible influence in Ero Dam. It will equally provide environmental awareness for both the public and government for necessary attention.

MATERIAL AND METHODS

Samples

Representative water samples were taken below the water surface using one litre acid leached polythene bottles at 12 different locations along Ero River (Fig. 1) for the period of dry and wet seasons. The samples were chemically preserved by the addition of 5ml concentrated HNO₃ for a litre of sample. Sediments samples were collected by divers at the points where the water samples were collected for both seasons (dry & wet) and stored in a polythene bag that has been previously washed and leached with 10% HNO₃.

Samples treatment

Five Cm³ of concentrated hydrochloric acid was added to 250mL of each of the water samples and evaporated to 25mL. The concentrate was transferred to 100ml volumetric flask and diluted to mark with deionized water (parker; 1972). 5g of each of the sediments samples was digested using Nwajei and Gagophien (2000) method. The resultant solutions from the digest were analyzed for heavy metals using atomic absorption spectroscopy (Buck model 200A). Result obtained were averages of replicate determinations.

RESULTS AND DISCUSSION

Tables 1 & 2 present the concentration (mg/l) of heavy metals in water sample for both wet and dry season for different location in Ero River. In the wet season, the concentration of Zn, Mn, Cu, Fe, Cr ranged between 0.70 – 0.82, 0.30 – 0.54, 0.10 – 0.22, 0.65 – 1.02, and 0.20 – 0.26 mg/l respectively while during the dry season, it ranged between 0.89 – 0.98, 0.30 – 0.42, 0.29 – 0.44, 0.37 – 0.62 and 0.27 – 0.48 mg/L respectively.

Table 1: Concentration mg/L of some heavy metals in water of ERO River; (Wet Seasons)

Location	Metals							
	Zn	Mn	Cu	Fe	Ni	Cd	Co	Cr
L1	0.72	0.30	0.10	0.65	ND	ND	ND	0.20
L2	0.72	0.35	0.15	0.80	ND	ND	ND	0.24
L3	0.70	0.42	0.12	0.90	ND	ND	ND	0.30
L4	0.74	0.40	0.20	1.02	ND	ND	ND	0.22
L5	0.73	0.34	0.10	0.80	ND	ND	ND	0.24
L6	0.70	0.36	0.15	0.60	ND	ND	ND	0.26
L7	0.79	0.52	0.18	0.70	ND	ND	ND	0.24
L8	0.78	0.36	0.20	0.65	ND	ND	ND	0.20
L9	0.70	0.54	0.15	1.08	ND	ND	ND	0.26
L10	0.82	0.50	0.14	1.02	ND	ND	ND	0.26
L11	0.78	0.32	0.18	0.70	ND	ND	ND	0.26
L12	0.80	0.50	0.22	0.72	ND	ND	ND	0.24
Mean	0.75	0.41	0.16	0.80	-	-	-	0.24
SD	0.04	0.08	0.04	0.16	-	-	-	0.03
CU	5.3	19.5	25.0	2.0	-	-	-	12.5
EX	9.0	4.9	1.9	9.64	-	-	-	2.9
EX ²	6.8	2.1	0.31	8.0	-	-	-	0.72

In most cases, the value of metal concentration indicates that Iron, Zinc, manganese are the most abundant metals in the water sample during the period of the present study. Cadmium was only detected in the water samples during the dry season (Nd.-0.6). From the tables it was observed that the concentration of metals varies between the two seasons. Of all the metals examined for both seasons in the water samples, Iron, Zinc and manganese have the highest values, similar observation have been reported for aquatic environment in Nigeria (Asaolu *et al.*, 1997; Kakulu and Osibanjo, 1988, Adefemi *et al.*, 2004, Adefemi *et al.*, 2008). The high concentration of Iron and zinc in water sample could be of advantage because it has been observed that this metal is an essential nutrients that are required in enzymatic biochemical activities in the body (Asaolu *et al.*, 1997, Kakulu and Osibanjo 1988, Adefemi *et al.*, 2004, Oshodi and Ipinmoroti, 1993). Except in Iron, the metal concentration values for dry season are relatively higher than those of the wet season; this might be due to the inflow of water and other particles during the raining season which might have diluted the concentration of the metals. This is similar to the

observation of Asaolu *et al.*, (1997) and Adefemi *et al.*, (2007).

Tables 3 and 4 presents the concentration (mg/g) of heavy metals in the sediment samples, all the metals examined were detected in the sediment samples from the river. The concentrations of Zn, Mn, Cu, Fe, Ni, Cd, Co, Cr ranged between 52.4 – 63.4, 117.2 – 140.1, 33.2 – 42.5, 29.2 – 340.0, 5.4 – 13.1, 780 – 98.2, 8.3 – 11.1 and 35.2 – 44.2 mg/g respectively during the wet season while for dry season, the concentration of the metals detected ranged from 42.3 – 53.3, 107.6 – 130.0, 33.2 – 37.7, 210.1 – 300.7, 6.0 – 12.7, 50.1 – 67.3, 7.2 – 10.1 and 25.9 – 37.6 mg/g respectively.

For the period of study, the concentrations of metals in dry season were higher than those of the wet season. This is similar to the observation of Awofolu (2005) who studied the determination and seasonal variation of heavy metals in algae and sediments in senders from industrial areas in Lagos State, Nigeria. For all the metals examined, iron was found to be consistently higher throughout the period of study for both seasons (dry & wet). Similar

Table 2: Concentration mg/l of some heavy metals in water of ERO River; (Dry Seasons)

Location	Metals							
	Zn	Mn	Cu	Fe	Ni	Cd	Co	Cr
L1	0.91	0.41	0.29	0.62	ND	ND	ND	0.48
L2	0.98	0.40	0.42	0.37	ND	0.60	ND	0.30
L3	0.96	0.38	0.34	0.50	ND	ND	ND	0.27
L4	0.93	0.39	0.34	0.44	ND	ND	ND	0.42
L5	0.92	0.42	0.34	0.50	ND	ND	ND	0.35
L6	0.89	0.41	0.40	0.40	ND	ND	ND	0.40
L7	0.90	0.36	0.44	0.38	ND	ND	ND	0.42
L8	0.94	0.36	0.42	0.45	ND	ND	ND	0.36
L9	0.91	0.38	0.40	0.38	ND	ND	ND	0.40
L10	0.93	0.40	0.34	0.50	ND	ND	ND	0.36
L11	0.92	0.30	0.32	0.40	ND	ND	ND	0.32
L12	0.96	0.40	0.36	0.60	ND	ND	ND	0.32
X	0.93	0.38	0.36	0.46	-	0.05	-	0.37
SD	0.03	0.03	0.05	0.09	-	0.17	-	0.06
CU	3.2	7.9	13.9	19.6	-	3.4	-	16.0
EX	11.2	4.6	4.4	5.5	-	0.6	-	4.4
EX ²	10.4	1.78	1.6	2.6	-	0.3	-	1.7

observation have been reported for iron in the Niger-Delta area (Kakulu, 1988), and also Asaolu (1997) for the Coastal area of Ondo State. Egila and Nimyel (2002) while studying the chemical speciation of

Table 3: Concentration (mg/kg) of some heavy metals in sediment of ERO River (Dry Seasons)

Location	Metals							
	Zn	Mn	Cu	Fe	Ni	Cd	Co	Cr
L1	63.4	139.9	36.5	305.8	13.1	90.3	10.8	44.2
L2	60.4	140.1	40.8	335.2	8.7	980.2	10.5	44.0
L3	56.4	120.1	41.2	340.0	5.4	74.5	11.1	40.3
L4	60.2	134.3	42.5	310.5	8.0	70.1	10.8	39.2
L5	52.4	118.5	34.2	300.2	7.0	81.1	9.0	40.0
L6	53.5	118.2	35.6	290.2	7.0	80.2	9.1	38.9
L7	55.7	117.2	35.1	298.0	7.0	80.0	9.1	40.1
L8	55.9	118.9	35.0	300.0	7.0	78.7	10.2	37.0
L9	56.0	118.7	34.7	320.1	7.0	79.3	10.1	35.2
L10	58.3	119.0	33.2	300.0	7.0	78.7	9.2	36.0
L11	56.0	117.7	34.5	300.0	7.1	78.0	8.7	37.2
L12	67.2	129.7	35.6	310.1	7.0	80.1	8.3	36.2
X	57.1	124.4	36.6	312.7	7.6	80.8	9.7	39.0
SD	3.06	9.0	3.1	21.0	1.9	7.2	0.95	2.9
CU (%)	5.4	7.2	8.5	6.7	25.0	8.9	9.8	7.4
EX	685.4	1492	439	3752	91.6	969	117	468
EX ²	39250	18643	16159	1178025	734	78846	1149	18369

Table 4: Concentration (mg/kg)/L of some heavy metals in sediment of ERO River (Wet Seasons)

Location	Metals							
	Zn	Mn	Cu	Fe	Ni	Cd	Co	Cr
L1	58.3	129.8	37.7	300.7	12.7	67.3	9.9	38.6
L2	50.3	130.0	36.8	282.1	9.6	63.3	9.4	37.0
L3	46.3	110.0	30.6	210.1	6.8	54.3	10.1	34.1
L4	50.1	124.2	34.2	240.1	7.9	50.1	9.8	36.0
L5	42.3	108.1	35.1	262.2	7.1	61.1	8.8	26.1
L6	43.4	109.2	34.2	262.0	7.0	60.1	8.8	26.1
L7	45.6	108.8	34.0	260.0	6.8	60.0	8.4	26.1
L8	45.8	108.6	33.2	261.2	6.8	58.7	9.0	25.9
L9	46.0	107.9	33.0	260.4	6.7	59.2	8.2	25.9
L10	46.2	107.6	34.1	258.2	6.7	58.7	8.2	26.1
L11	45.9	119.2	34	258.0	6.7	58.0	7.9	25.8
L12	46.0	109.7	33.2	240.1	6.0	61.1	7.2	25.4
X	64.8	114.4	34.2	257.9	7.6	59.3	8.8	29.3
SD	3.0	8.9	1.8	22.1	1.8	4.3	0.88	5.1
CU (%)	6.4	7.8	5.3	8.5	23.7	7.3	10.0	17.4
EX	561.2	1373	4.10	3095	90.8	712	106	352
EX ²	26347	157978	14051	803696	724	42433	940	10619

nearly even distribution of the metals in the aquatic system. This could indicate that the source of metals levels in the system is from a non point source since the river flows through the forest.

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