The water quality assessment at historical places, Ellora caves and Khultabad area of Aurangabad region

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ABSTRACT

Assessment of drinking water quality which involves physicochemical parameters revealed that the sample characteristics are varied considerably. In all, nine water samples were selected, out of which four samples were from Ellora caves and five from Khultabad. The water samples were assessed for parameters such as Temperature, pH, BOD, COD, Hardness, SO_4^{-2} , Nitrate and Fluoride. The period of assessment was from September to December 2008.

Key words: Physicochemical parameters, Well water, Bore-well water, Tank.

INTRODUCTION

With its many fold shining aspects, Aurangabad is also world famous for its historical Ellora caves and Mughal emperor Aurangajeb's Tomb at Khultabad. These places are 33 and 30 km. away respectively from Aurangabad. The ancient town Ellora is situated on the bank of river Yelganga which originates from nearby basaltic mountains of Deccan platue. The location oif Ellora is 20° 00' N and 75° 10' E. It is also famous for its pilgrim the Ghrishneshwar Temple, one of twelfth Jotirlinga of lord Shiva. The Vishwakarma Shivalaya Kund which is also called as Ahilyabai Holkar Tank and Janardhan Swami Matha are another specialties of Ellora. Thousands of people from every corner of world visit Ellora. Therefore it was proposed to assess the water quality of this area.

Four sampling sites are from Ellora as Shivalaya Kund, Ghrishneshwar Temple well, well from Janardhan Swami matha and bore well sample from Pawan Ganpati mandir area. Another sampling station selected was Khultabad, situated to south east of Ellora. The mughal emperor Aurangajeb's Tomb and Bhadramaruti mandir are in Khultabad. The famous hill stations "Mhaismal" and "Shulibhanjan" are near by places. Five sampling sites were selected from Khultabad area, out of them "Dharam Talab" and "Pangra Talab" and parionka talab were tanks. One sample was from well at "Bhadramaruti Temple" and the other was from bore well situated at the centre of town.

MATERIAL AND METHODS

All samples were collected in 500 mL sterilized bottles for physicochemical analysis. AR grade chemicals were used for preparation of regents in double distilled glass water. Physical parameters like Temperature, pH were measured on the spot by using water analysis kit. Remaining parameters were determined by using standard procedures¹.

RESULTS AND DISCUSSION

The results on water quality from different sampling sites are given in Table 1 and 2 and are interpreted subsequently.

- S1 = Tank (Ahilyabai Holkar Tank)
- S2 = Dug Well (Ghrishneshwar Tample area)
- S3 = Dug Well (Janardhan Swami Math area)
- S4 = Bore Well (Pawan Ganesh Mandir area)
- S5 = Tank (Dharam Talab)
- S6 = Tank (Pangra Talab)
- S7 = Dug Well (Bhadra Maruti Temple area)
- S8 = Bore Well (Khultabad Town)
- S9 = Tank (Parionka Talab)

Temperature

The various chemical and biological actions in water organisms depend upon temperature². The temperature of water depends on atmospheric conditions. The increase in temperature of water reduces solubility of gases, changes the taste and fastens the chemical reactions in water³. The maximum temperature 26.2 °C was recorded at sit S9 and minimum 24.1 °C at sit S4.

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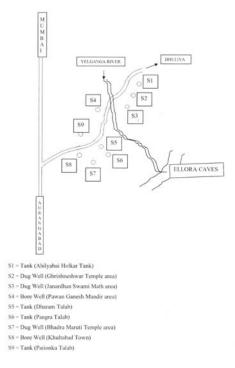
The pH is one of the most important parameters deciding the extent of water pollution⁴. It affects directly the chemical and biological reactions in water⁵. In present study the pH range noted was nearly neutral i.e. 6.90 to 7.25. The maximum pH value 7.40 was noted at site S5. The pH range given by the ISI standards⁶ for bathing and drinking water is in between 6.50 to 8.50⁷. Hence it can be concluded that the pH range of these samples are within range.

BOD

The BOD is one of the most important parameters in water quality assessment⁸. It reflects in physical, chemical and biological processes prevailing in water. Its maintenance is very important for survival of biological forms. The permissible limit of BOD is 8 ppm⁹. In the present investigations the values of BOD are in desired limit.

COD:

The COD plays important role in deciding the extent of water pollution. The permissible limit



of COD is 100-150 ppm. High COD values have detrimental effects on aquatic life. In the present investigations the values of COD are in acceptable limit.

Hardness

The total hardness is an important parameter of water quality assessment. The hardness of water is due to salts of calcium, magnesium with carbonates, bicarbonates and sulphates¹⁰. The degree of hardness can be classified¹¹ as:

ppm	water quality
<50	soft
50-150	moderately hard
150-300	hard
>300	very hard

The total hardness of water samples under investigation ranges from 78 to 275 ppm.

Chlorides

The chloride is the best indicator of water pollution¹². The chloride concentration in water is affected by evaporation and precipitation ^{13, 14}. The permissible limit of chloride concentration is 250 ppm. The amount of chloride in all water samples

				Table 1: H	ydrochemic	al analysis	Table 1: Hydrochemical analysis of water samples	mples			
S. No.	Site No.	Source	Temp °C	Ηd	BOD	COD	Hardness	Chloride	Sulphate	Nitrate	Fluoride
-	S1	Tank	25.0	6.19	3.6	36.0	125.3	90.74	190.2	8.20	0.62
7	S2	D.W.	24.8	7.10	2.8	25.8	78.0	154.9	108.4	4.30	0.72
с	S3	D.W.	25.2	7.00	2.7	23.2	86.4	134.9	102.2	4.45	0.80
4	S4	B.W.	24.1	7.20	1.9	18.9	258.0	160.6	139.3	1.91	0.95
5	S5	Tank	25.3	7.25	3.2	48.5	185.0	124.6	200.1	6.60	06.0
9	S6	Tank	25.7	70.00	3.2	48.4	175.0	134.8	210.6	5.20	1.00
7	S7	D.W.	25.4	7.10	1.4	14.9	168.2	164.9	115.0	3.20	0.32
œ	S8	B.W.	25.4	7.20	1.2	16.8	275.0	154.9	200.3	1.60	0.78
6	S9	Tank	26.2	7.15	3.1	38.2	209.2	105.0	189.1	6.30	0.98
All the I D. W. =	All the parameters a D. W. = Dug Well	All the parameters are expressed in ppr D. W. = Dug Well	d in ppm exce B. Table 2	m except pH and Temperature B.W. = Bore Well able 2: The maximum & min	emperature ell um & minin	num value	m except pH and Temperature B.W. = Bore Well B.M. = Bore Well	parameters			
Parameter		Temp °C	Hd	BOD	COD	Hardness		Chloride Sulphate	late Nitrate		Fluoride
Minimum		24.1	6.91	1.20	14.9	78.00		90.74 102.2	.2 1.60		0.32
Maximum		26.2	7.25	3.60	45.8	258.0		164.9 21.6	6 8.20		1.00

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water
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205

0.81

4.90

156.4

117.8

168.0

31.7

2.40

7.08

25.12

Average

ranges from 90.0 ppm to 164.9 ppm and do not crosses the suggested limits.

Sulphates

Sulphates generally have less effect on taste than chlorides and carbonates. The recommended upper limit of sulphates ions for human consumption is 250 ppm¹⁵. The highest concentration of sulphates is found to be 210 ppm at site S6. The concentration of sulphates ranges from 102.2 to 210 ppm. The values are in accordance with ISI limits.

Nitrates

The most important source of nitrate is biological oxidation of organic nitrogenous

substances. The agricultural run off is also a source of nitrate. In present investigations the high values of nitrates 8.2 ppm are recorded at site S1 and minimum values 1.60 ppm were at site S8, which are within range.

Fluoride

Fluoride is beneficial up to certain extent, when present in concentration range of 0.8 - 1.0 ppm for calcification of dental enamel specifically for children below eight years¹⁶. High levels of fluoride can cause dental fluorosis (Fluoride > 1.5 ppm) and skeletal fluorosis (Fluoride > 3.0 ppm). In present investigation the fluoride concentration is found to be in desired limit.

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