

Assessment of water quality near Chopan River, Raghogarh, District Guna, (M.P.)

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ABSTRACT

The study of the contamination level of drinking water used and effect of fertilizer industry of aquatic biota in Raghogarh was conducted at 6 stations in Chopan River and 4 stations of surrounding tube wells in year (2004).

In the water samples, temperature, pH, conductivity, TDS, hardness, calcium, Magnesium, DO chloride, free ammonia-N were evaluated through analysis. It is found that the values of total dissolved solids and free ammonia-N are higher. Total dissolved solids in excessive causes laxative effect, whereas increase in ammonia cause decrease in pH ultimately it is harmful to life, irrigation and drinking purpose.

Key words: Water quality, Chopan river, Raghogarh.

INTRODUCTION

Water is one of the most important commodities which man has exploited than any other resources for the sustenance of his life. Most of our demands for water is fulfilled by rainwater which gets deposited in surface and ground water resources. Though, water is continuously purified by evaporation and precipitations, pollution of water has emerged as one of the most significant environmental problems of recent times. The cause of such a situation is many but the quality of water is deteriorated day by day due to rapid increase in urbanization and industrialization¹.

Some of the major adverse effect have arisen from construction of reservoirs, surface water irrigation, deforestation, industrial and urban waste disposal.

Water quality deterioration has made potable water resources scarcer and endangered for plant and animal life². In the present work, various water samples of tube wells and river water of Raghogarh Tehsil has been analyzed.

The impact of fertilizer industry on different stations at the factory is engaged in the production of urea i.e. nitrogenous fertilizer from natural gas. The wastes from fertilizer industries can also be a rich source of nutrients causing eutrophication³.

Many authors⁴⁻⁸ have studied the physicochemical characterization of groundwater and surface water of different parts of the world. The objective of present study was to determine the quality of water supplied to the rural community in order to estimate the health implications.

EXPERIMENTAL

Water samples were collected for three seasons i.e., summer, rainy and winter. Sample within 10 kms. from fertilizer industries were collected from ten stations, out of these 10, four are tubewells and six stations are on Chopan River. Water samples were collected in polythene bags from various stations. Analyses for physicochemical parameters were done by following methods described in the literature⁹⁻¹⁰. Hardness is determined by EDTA-titrimetric methods, chloride

by Mohr's methods, DO by Winkler's method and free ammonia by Nesslerization method.

RESULTS AND DISCUSSION

All the metabolic and physiological activities and life processes of aquatic organisms are generally influenced by temperature. It accelerates chemical reactions.

In the present study the temperature in tube well is ranged from 17°C to 31°C and in river water it was noticed in the range of 18.5°C to 32.5°C. The temperature was recorded at the sites only.

The pH value of drinking water is an important index of acidity, alkalinity and resulting value of acid base interaction of no. of its minerals and organic components. During the study period pH between 7.4 to 8.3 was noticed in tube wells and in river water it was ranged from 7.3-8.35. Specific conductivity of water provides quick and convenient method for determination of total amount of ionizable salts present in it and expressed as umhos/cm. In the present finding specific conductivity in tube wells varied from 170 to 680 umhos/cm and in river water it is 240 to 520 umhos/cm. Higher conductivity is observed at station R₃ after confluence of waste water from fertilizer factory in summer and low in winter seasons. According to world health organization and Indian standard, total dissolved solids values should be lower than 500 mg/L for drinking water. In the present study total dissolved solids range from 170 to 659 mg/L in tube well and 160 to 585 mg/L in Chopan River. Highest TDS was noticed as Station R₃ most of the samples have higher values of TDS as prescribed by WHO and Indian standards.

Total hardness is a measure of the capacity of precipitate soap. It is the sum of polyvalent cation present in water. It was found to have been varied from 150 to 518 mg/L in tube wells and 133 to 296 mg/L was observed in river water. Samples values as compared to IST and WHO.

Calcium is needed for the body in small quantities. Calcium hardness is mainly due to salts present in water and expressed as calcium

Table - 1: Physico Chemical characteristic of water of different tubewells (2004)

S.No.	Sites	Season	Temp.	pH	Conductivity	TDS	Hardness	Parameters			Free Ammonia-N
								Ca	Mg	Cl	
1	TW1	Summer	25	7.6	260	480	392	250	142	58	0.4
		Rainy	17	7.7	190	424	298	194	104	30	0.2
		Winter	18	7.7	310	602	476	304	170	98	0.66
2	TW2	Summer	31	7.9	440	518	312	202	110	100	0.6
		Rainy	26.5	7.5	360	436	290	180	82	54	0.2
		Winter	20	7.8	170	352	224	160	64	30	Nil
3	TW3	Summer	29	8	360	659	518	218	300	90	0.6
		Rainy	26	7.5	270	598	472	200	272	58	0.21
		Winter	20	7.5	200	466	458	201	257	25	0.14
4	TW4	Summer	31	8.3	680	596	252	180	72	96	0.65
		Rainy	24	7.4	380	450	182	142	40	54	0.14
		Winter	19	7.6	250	366	158	120	38	36	0.18

Table - 2: Physico-chemical characteristics of river Chopan at different stations (2004)

S.No.	Sites	Season	Temp.	pH	Conductivity	TDS	Hardness	Parameters				
								Ca	Mg	DO	Cl	Free Ammonia-N
1	R1	Summer	31	8.3	460	382	226	132	94	7	30	0.3
		Rainy	26.5	7.4	350	240	170	90	80	7.4	26	0.12
		Winter	18.5	7.9	270	162	185	105	80	10.8	15	0.18
2	R2	Summer	31	8.35	500	390	248	150	98	6.9	32	0.56
		Rainy	26	7.3	360	242	172	90	80	7.2	26	0.14
		Winter	18	8	300	210	204	110	94	11.2	17	0.24
3	R3	Summer	31.5	7.63	520	468	296	184	112	6.6	46	0.78
		Rainy	27	7.4	360	246	170	92	82	7.4	25	0.16
		Winter	19	7.9	330	350	232	148	84	8.8	23	0.45
4	R4	Summer	31	7.7	440	450	270	156	114	6.8	42	0.45
		Rainy	25	7.4	350	240	175	90	118	7.4	25	0.14
		Winter	19	7.9	250	322	224	126	98	9.2	23	0.35
5	R5	Summer	30.5	7.65	460	460	285	165	120	6.5	44	0.65
		Rainy	26	7.3	360	240	170	92	82	7.4	22	0.12
		Winter	19	8	270	348	226	134	92	8.8	25	0.35
6	R6	Summer	30.5	7.7	350	430	266	160	106	7	37	0.6
		Rainy	26	7.3	350	246	170	90	80	8.4	25	0.12
		Winter	19	8.1	240	315	133	133	95	9.8	24	0.42

carbonates. The limits of calcium and magnesium have been prescribed in the range of 75 to 200 mg/L and 50 to 100 mg/L respectively. But the observed calcium values were higher i.e, 120 to 304 mg/L in tube wells and 90 to 200 mg/L in river water. Magnesium is in the range of 38 to 300 mg/L in tube well water and 80-112 mg/L in river water.

Presence of chloride in high amount in water indicates that it is contaminated by sewage. But in the present study quantity of chloride is moderate, i.e. in the Chopan river its is ranged from 30 to 100 mg/L in tube well and 15 to 46 mg/L in river water. The desirable limit of chlorides is 250 mg/L and permissible up to 600 mg/L.

DO is one of the most important constituents of water which effect physiological and biological process. In the present work the observed DO is 0.3 to 7.8 mg/L in tube wells and in river water it ways ranged from 6.5 to 11.2 mg/L. Free ammonia-N was measured by using Nessler's reagent. The colour was measured by using

spectrophotometer. In the present study. It was ranged from 0.14 mg/L to 0.66 mg/L in tube wells and in river water it is ranged from 0.12 to 0.78mg/L.

Conclusion

It is seen that the sample near the fertilizers have higher values particularly total dissolved solids and free-ammonia-N. High level of TDS causes objectionable taste and laxative effect. The increase in free ammonia causes organic pollution. Both these contents in water in increased amount are harmful to fish and other biota. So it is necessary to control the incoming industrial wastage by redesigning the infrastructure to protect the river and surroundings.

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